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(54) Title: PHOTO-CURABLE RESIN COMPOSITION			
(57) Abstract			
<p>A novel photo-curable resin composition. The resin composition capable of providing a three-dimensional object having excellent mechanical strength and high dimensional accuracy. The resin composition capable of providing a three-dimensional object with excellent stability in shape and properties using an optical fabricating process. The composition comprises (A) a cationic polymerized compound, (B) a cationic photo-initiator, (C) an ethylenically unsaturated monomer, (D) a radical photo-initiator, (E) a polyether polyol having on average about 3 or more hydroxyl groups and wherein at least 60 % by weight of the ethylenically unsaturated monomer having three or more ethylenically unsaturated bonds in one molecule.</p>			
<p><i>additives p 19 polybutadiene 5-B-5 block copoly.</i> <i>p 20 resin particles</i> <i>no mention of core-shell polymers</i></p>			

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TITLE OF THE INVENTION  
PHOTO-CURABLE RESIN COMPOSITION

5

BACKGROUND OF THE INVENTION

Field of the Invention

10

The present invention relates to a photo-curable resin composition, especially to a liquid photo-curable resin composition suitable as material used in photo-fabrication processes.

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Description of the Background Art

In recent years, photo-fabrication processes for forming three-dimensional objects consisting of integrally laminated cured resin layers prepared by repeating a step of forming a cured resin layer by selectively irradiating a liquid photo-curable material with light have been proposed (see Japanese Patent Application Laid-open No. 247515/1985, USP No. 20 4,575,330 (Japanese Patent Application Laid-open No. 35966/1987), Japanese Patent Application Laid-open No. 101408/1987, Japanese Patent Application Laid-open No. 24119/1993).

A typical example of such a photo-fabrication process comprises forming a curable resin layer having a specified pattern by selectively irradiating with light using, for example, an ultraviolet radiation laser on the surface of a liquid photo-curable material (photo-curable resin composition), feeding the 30 photo-curable resin composition equivalent to one layer to form another thin layer of the composition over this cured resin layer, and selectively irradiating this

thin layer with light to form a new cured resin layer which is integrally laminated over the previously formed cured resin layer. This step is repeated a number of times, with or without changing the pattern in which the light is irradiated to form a three-dimensional object consisting of integrally laminated multiple cured resin layers. This photo-fabrication process has been attracting considerable attention, because the target three-dimensional object can easily be prepared in a short period of time even if it has a complicated shape.

The following resin compositions (A) to (C) are reported as the photo-curable resin composition used in the photo-fabrication process.

(A) Resin compositions containing a radical polymerizable organic compound such as urethane(meth)acrylate, oligoester(meth)acrylate, epoxy(meth)acrylate, thiol-ene compounds, photosensitive polyimide, and the like (see Japanese Patent Applications Laid-open No. 204915/1989, No. 208305/1990, and No. 160013/1991).

(B) Resin compositions containing a cationic polymerizable organic compound such as an epoxy compound, cyclic ether compound, cyclic lactone compound, cyclic acetal compound, cyclic thioether compound, spiro-orthoester compound, vinyl ether compound, and the like (see Japanese Patent Application Laid-open No. 213304/1989).

(C) Resin compositions containing a radical polymerizable organic compound and a cationic polymerizable organic compound (see Japanese Patent Applications Laid-open No. 28261/1990, No. 75618/1990, and No. 228413/1994).

The characteristics required of the photo-curable resin composition used for these photo-fabrication processes include a low viscosity to

form a smooth liquid level and the capability of being rapidly cured by irradiation of light. Also, the required characteristics of the photo-curable resin composition are minimal swelling of the cured products and minimal deformation due to shrinkage during curing with light, such as production of warped parts, indented parts (sinkmark), or stretched parts (overhanging parts).

Three-dimensional objects prepared by photo-fabrication methods have conventionally been used for design model and trial mechanical parts. High dimensional accordance with the design model in fine processing, mechanical strength and heat resistance sufficient to withstand conditions of use are especially demanded of these trial mechanical parts. However, no conventional resin composition can satisfy the above demands. The three-dimensional objects obtained exhibit problems of deformation with the passage of time, such as production of warped parts, or indented parts, or stretched parts (overhanging parts), because of residual strain due to the shrinkage during curing. These problems of deformation with the passage of time can be partly solved by the correction of the input data to the CAD. However, the correction is insufficient to correspond to recent progress in fineness and complication in shape, and to circumstantial variations of use.

Mechanical characteristics of three-dimensional objects prepared by photo-fabrication by using conventional resin compositions containing an epoxy compound are degraded in the course of time in accordance with environmental conditions such as temperature and humidity. Therefore, use of these three-dimensional objects is unsuitable for the application requiring mechanical strength over an extended period of time.

Also, no three-dimensional object prepared by

using conventional resin compositions is provided with mechanical strength, dimensional accuracy, heat resistance, and the like.

The present invention has been achieved in view of this situation and has an object of providing a novel photo-curable resin composition.

Another object of the present invention is to provide a photo-curable resin composition providing three-dimensional objects which have mechanical strength and high dimensional accuracy and which is suitable for trial mechanical parts.

Further object of the present invention is to provide a photo-curable resin composition which can be fabricated to produce a three-dimensional object provided with lasting minimal deformation characteristics.

A still further object of the present invention is to provide a photo-curable resin composition which can be fabricated to produce a three-dimensional object provided with lasting minimal degradation in mechanical characteristics.

#### SUMMARY OF THE INVENTION

The above first object can be attained in the present invention by a photo-curable resin composition comprising,

- (A) a cationic polymerizable organic compound,
- (B) a cationic photo-initiator,
- (C) an ethylenically unsaturated monomer,
- (D) a radical photo-initiator, and
- (E) a polyether polyol having three or more hydroxyl groups, wherein at least 60% by weight of the said ethylenically unsaturated monomer (C) is a polyfunctional monomer having three or more ethylenically unsaturated bonds in one molecule.

DETAILED DESCRIPTION OF THE INVENTION  
AND PREFERRED EMBODIMENTS

The present invention will now be explained  
5 in detail.

<Cationic polymerizable organic compound>

The cationic polymerizable organic compound  
(hereinafter called [component (A)] ) contained in the  
10 photo-curable resin composition of the present  
invention is an organic compound which can polymerize  
or crosslink by photo-irradiation by irradiating light  
in the presence of a cationic photoinitiator. Typical  
examples of the component (A) are epoxy compounds,  
15 oxetane compounds, oxolane compounds, cyclic acetal  
compounds, cyclic lactone compounds, thiirane  
compounds, thietane compounds, vinyl ether compounds,  
spiro-ortho ester compounds which are reaction products  
of an epoxy compound and lactone, ethylenically  
20 unsaturated compounds, cyclic ether compounds, cyclic  
thioether compounds, and vinyl compounds.

For example, the following compounds can be  
illustrated as the epoxy compound which can be used as  
the component (A): bisphenol A diglycidyl ether,  
25 bisphenol F diglycidyl ether, bisphenol S diglycidyl  
ether, brominated bisphenol A diglycidyl ether,  
brominated bisphenol F diglycidyl ether, brominated  
bisphenol S diglycidyl ether, an epoxy novolac resin,  
hydrogenated bisphenol A diglycidyl ether, hydrogenated  
30 bisphenol F diglycidyl ether, hydrogenated bisphenol S  
diglycidyl ether, 3,4-epoxycyclohexylmethyl-  
3',4'-epoxy-cyclohexane carboxylate, 2-(3,4-epoxycyclo-  
hexyl-5,5-spiro-3,4-epoxy)cyclohexane-metha-dioxane,  
bis(3,4-epoxy-cyclohexylmethyl)adipate,  
35 vinylcyclohexene oxide, 4-vinyl epoxycyclohexane,  
bis(3,4-epoxy-6-methylcyclohexylmethyl) adipate,  
3,4-epoxy-6-methyl cyclohexyl-3',4'-epoxy-6'-

- methyl-cyclohexane carboxylate, methylenebis(3,4-epoxy-cyclohexane), dicyclopentadiene diepoxide,  
di(3,4-epoxy-cyclohexylmethyl) ether of ethylene  
glycol, ethylene- bis(3,4-epoxycyclohexanecarboxylate),  
5 epoxyhexahydrodioctyl phthalate, di-2-ethylhexyl  
epoxyhexahydrophthalate, 1,4-butanediol diglycidyl  
ether, 1,6-hexanediol diglycidyl ether, glycerol  
triglycidyl ether, trimethylolpropane triglycidyl  
ether, polyethylene glycol diglycidyl ether,  
10 polypropylene glycol diglycidyl ethers; polyglycidyl  
ethers of polyether polyol obtained by adding one or  
more of alkylene oxide to aliphatic polyhydric alcohol  
such as ethylene glycol, propylene glycol, or glycerol;  
diglycidyl esters of aliphatic long chain dibasic acid;  
15 mono glycidyl ethers of aliphatic higher alcohol; mono  
glycidyl ethers of phenol, cresol, butyl phenol, or  
polyether alcohol obtained by addition of alkylene  
oxide to phenol, cresol, or butyl phenol; glycidyl  
esters of higher fatty acid; epoxidated soybean oil,  
20 butyl epoxystearate, octyl epoxystearate, epoxidated  
linseed oil, and epoxidated polybutadiene.  
Other compounds which can be used as the component (A)  
are oxetanes such as trimethylene oxide, 3,3-dimethyl  
oxetane, 3,3-dichloromethyl oxetane, 3-ethyl-3-  
25 phenoxymethyl oxetane, and bis(3-ethyl-3-methyloxy)  
butane; oxolane such as tetrahydrofuran and  
2,3-dimethyltetrahydrofuran; cyclic acetals such as  
trioxane, 1,3-dioxolane, and 1,3,6-trioxan cyclooctane;  
cyclic lactones such as  $\beta$ -propiolactone and  
30  $\epsilon$ -caprolactone; thiirane such as ethylene sulfide,  
1,2-propylene sulfide, and thioepychlorohydrin;  
thiethanes such as 3,3-dimethyl thiethane; vinyl ethers  
such as ethylene glycol divinyl ether, triethylene  
glycol divinyl ether, and trimethylolpropane trivinyl  
35 ether; spiro-ortho esters which are obtained by a  
reaction of epoxy compound and lactone; ethylenically  
unsaturated compounds such as vinyl cyclohexane,



isobutylene, and polybutadiene; and their derivatives.

Among these cationic polymerizable compounds, desirable compounds are bisphenol A diglycidyl ether, bisphenol F diglycidyl ether, hydrogenated bisphenol A diglycidyl ether, hydrogenated bisphenol F diglycidyl ether, 3,4-epoxy-cyclohexylmethyl-3'4'-epoxycyclohexane carboxylate, bis-(3,4-epoxycyclohexylmethyl) adipate, 1,4-butanediol diglycidyl ether, 1,6-hexanediol diglycidyl ether, glycerol triglycidyl ether, trimethylolpropane triglycidyl ether, polyethylene glycol diglycidyl ether, and polypropylene glycol diglycidyl ether.

Particularly preferred cationic polymerizable compounds as the component (A) are the compounds possessing two or more alicyclic epoxy groups in a molecule, such as 3,4-epoxy-cyclohexylmethyl-3'4'-epoxycyclohexane carboxylate or bis-(3,4-epoxycyclohexylmethyl) adipate. When this type of epoxy compounds are contained in an amount of 50% by weight or more in the component (A), the resulting resin composition has a large cationic polymerization rate (curing rate), so that not only the molding time can be reduced, but also curing shrinkage can be decreased. This ensures three dimensional objects exhibiting only small deformation over time.

The following commercially available products are desirable as the cationic polymerizable organic compound used as the component (A): UVR-6100, UVR-6105, UVR-6110, UVR-6128, UVR-6200, UVR-6216 (manufactured by Union Carbide Corp), Celoxide 2021, Celoxide 2021P, Celoxide 2081, Celoxide 2083, Celoxide 2085, Celoxide 2000, Celoxide 3000, Glycidole, AOEX24, Cyclomer A200, Cyclomer M100, Epolead GT-300, Epolead GT-301, Epolead GT-302, Epolead GT-400, Epolead 401, Epolead 403 (manufactured by Daicel Chemical Industries Co., Ltd.), Epicoat 828, Epicoat 812, Epicoat 1031, Epicoat 872, Epicoat CT508 (manufactured by Yuka Shell Co., Ltd.),

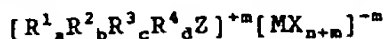
KRM-2100, KRM-2110, KRM-2199, KRM-2400, KRM-2410,  
 KRM-2408, KRM-2490, KRM-2200, KRM-2720, KRM-2750  
 (manufactured by Asahi Denka Kogyo Co., Ltd.),  
 Rapi-Cure DVE-3, CHVE, PEPC (manufactured by ISP Co.,  
 5 Ltd.), and VECTOMER 2010, 2020, 4010, 4020  
 (manufactured by Allied Signal Corp.).

The above cationic polymerizable compounds  
 may be used either individually or in combinations of  
 two or more.

10 The proportion of the component (A) in the  
 photo-curable resin composition of the present  
 invention is about 30-85% by weight, preferably about  
 40-80% by weight, and more preferably about 50-75% by  
 weight. If the proportion of the component (A) is too  
 15 low, the dimensional accuracy of the three-dimensional  
 object from the resin composition is decreased and the  
 three-dimensional object tends to deform with time. On  
 the other hand, if the proportion of the component (A)  
 is too high, the photo-curing characteristic of the  
 20 resulting resin composition is decreased, leading to a  
 reduction in the fabricating efficiency.

#### <Cationic photo-initiator>

The cationic photo-initiator (hereinafter  
 25 called [component (B)]) contained in the photo-curable  
 resin composition of the present invention is a  
 compound capable of generating a molecule initiating  
 cationic polymerization of the component (A) by  
 receiving radiation such as light. Given as especially  
 30 preferred examples of the cationic photo-initiator are  
 onium salts illustrated by the following formula, which  
 are compounds releasing Lewis acid on receiving light:



wherein the cation is onium; Z represents S, Se, Te, P,  
 35 As, Sb, Bi, O, I, Br, Cl, or N=N; R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, and R<sup>4</sup>  
 represent individually the same or a different organic  
 group; a, b, c, and d represent individually an integer

from 0 to 3, and the summarized number of a, b, c, and d ( $=a+b+c+d$ ) is equal to the valence number of Z. M represents a metal or metalloid which constitutes a center atom of a halide complex. Typical examples of M are B, P, As, Sb, Fe, Sn, Bi, Al, Ca, In, Ti, Zn, Sc, V, Cr, Mn, and Co. X represents a halogen atom. m is a substantial electric charge and n is the number of atoms in the halide complex ion.

Given as specific examples of the negative ion ( $MX_{n+m}$ ) in the above formula (1) are tetrafluoroborate ( $BF_4^-$ ), hexafluorophosphate ( $PF_6^-$ ), hexafluoroantimonate ( $SbF_6^-$ ), hexafluoroarsenate ( $AsF_6^-$ ), and hexachloroantimonate ( $SbCl_6^-$ ).

Also, onium salts represented by the general formula  $[MX_n(OH)]$  can be used. Further, onium salts including a negative ion, for example, perchloric acid ion ( $ClO_4^-$ ), trifluoromethane sulfonate ion ( $CF_3SO_3^-$ ), fluorosulfonate ion ( $FSO_3^-$ ), toluene sulfonate ion, trinitrobenzene sulfonate negative ion, and trinitrotoluene sulfonate negative ion are suitable.

Among these onium salts, especially effective onium salts are aromatic onium salts, in which the following compounds are preferable: aromatic halonium salts described, for example, in Japanese Patent Applications Laid-open No. 151996/1975 and No. 158680/1975; VIA aromatic onium salts described, for example, in Japanese Patent Applications Laid-open No. 151997/1975, 30899/1977, No. 55420/1981, and No. 125105/1980; VA aromatic onium salts described, for example, in Japanese Patent Application Laid-open No. 158698/1975; oxosulfoxonium salts described, for example, in Japanese Patent Applications Laid-open No. 8428/1981, No. 149402/1981, and No. 192429/1982; aromatic diazonium salts described, for example, in Japanese Patent Application Laid-open No. 17040/1974; and thiobililium salts described in the specification of USP No. 4,139,655. Iron/allene complex and aluminium

complex/photo-decomposed silica compound initiator are also given as examples of the onium salts.

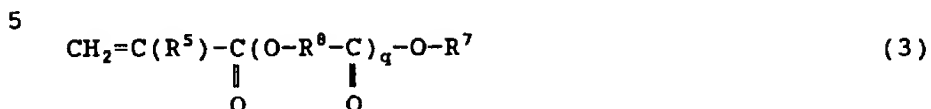
Given as preferred examples of commercially available products of the cationic photo-initiator which can be used as the component (B) are UVI-6950, UVI-6970, UVI-6974, UVI-6990 (manufactured by Union Carbide Corp), Adekaoptomer SP-150, SP-151, SP-170, SP-171 (manufactured by Asahi Denka Kogyo Co., Ltd.), Irgacure 261 (manufactured by Ciba Geigy), CI-2481, CI-2624, CI-2639, CI-2064 (manufactured by Nippon Soda Co., Ltd.), CD-1010, CD-1011, CD-1012 (manufactured by Sartomer Co., Ltd.), DTS-102, DTS-103, NAT-103, NDS-103, TPS-103, MDS-103, MPI-103, BBI-103 (manufactured by Midori Chemical Co., Ltd.). Among these, UVI-6970, UVI-6974, Adekaoptomer SP-170, SP-171, CD-1012, MPI-103 are particularly preferable to develop high curing sensitivity of the resin composition which contains these compounds.

The above-mentioned cationic photo-initiators can be used as the component (B) either individually or in combination of two or more.

The proportion of the component (B) in the photo-curable resin composition is about 0.1-10% by weight, preferably about 0.2-5% by weight, and more preferably about 0.3-3% by weight. If the proportion of the component (B) is too low, the photo-curing characteristic of the resin composition obtained is insufficient. No three-dimensional object having sufficient mechanical strength can be produced from this resin composition. On the other hand, if the proportion of the component (B) is too high, an appropriate light capability (curing depth) cannot be obtained when the resulting resin composition is used in the photo-fabrication process. Mechanical strength such as the toughness of the three-dimensional object prepared from this resin composition tends to be reduced.

## &lt;Ethylenically unsaturated monomer&gt;

- The ethylenically unsaturated monomer (hereinafter called [component (C)]) is a compound having ethylenically unsaturated groups (C=C) in the molecule. Given as typical examples of the component (C) are mono-functional monomers having one ethylenically unsaturated bond in one molecule, and polyfunctional monomers having two or more ethylenically unsaturated bonds in one molecule.
- Given as examples of the mono-functional monomer used as the component (C) are acrylamide, (meth)acryloyl morpholine, 7-amino-3,7-dimethyloctyl (meth)acrylate, isobutoxymethyl (meth)acrylamide, isobornyloxyethyl (meth)acrylate, isobornyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, ethyldiethylene glycol (meth)acrylate, t-octyl (meth)acrylamide, diacetone (meth)acrylamide, dimethylaminoethyl (meth)acrylate, diethylaminoethyl (meth)acrylate, lauryl (meth)acrylate, dicyclopentadiene (meth)acrylate, dicyclopentenyl (meth)acrylate, dicyclopentenyl (meth)acrylate, N,N-dimethyl (meth)acrylamide tetrachlorophenyl (meth)acrylate, 2-tetrachlorophenoxyethyl (meth)acrylate, tetrahydrofurfuryl (meth)acrylate, tetrabromophenyl (meth)acrylate, 2-tetrabromophenoxyethyl (meth)acrylate, 2-trichlorophenoxyethyl (meth)acrylate, tribromophenyl (meth)acrylate, 2-tribromophenoxyethyl (meth)acrylate, 2-hydroxyethyl (meth)acrylate, 2-hydroxypropyl (meth)acrylate, vinyl caprolactam, N-vinyl pyrrolidone, phenoxyethyl (meth)acrylate, butoxyethyl (meth)acrylate, pentachlorophenyl (meth)acrylate, pentabromophenyl (meth)acrylate, polyethylene glycol mono-(meth)acrylate, polypropylene glycol mono-(meth)acrylate, bornyl (meth)acrylate, methyltriethylene diglycol (meth)acrylate, and, the compounds shown in the following general formula (2) or (3).



10

wherein  $\text{R}^5$  represents a hydrogen atom or a methyl group,  $\text{R}^6$  represents an alkylene group containing 2-6, preferably 2-4 carbon atoms,  $\text{R}^7$  represents a hydrogen atom or an alkyl group containing 1-20, preferably 1-12 carbon atoms in which  $\text{R}^7$  may comprise ether groups like in a dioxane or tetrahydrofuran group; or  $\text{R}^7$  denotes an aromatic group optionally substituted with a  $\text{C}_1$ - $\text{C}_{12}$ , alkyl group, preferably a  $\text{C}_6$ -, alkyl group,  $\text{R}^8$  represents an alkylene group containing 2-8, preferably 2-5 carbon atoms,  $r$  denotes an integer from 0-12, preferably from 1-8, and  $q$  denotes an integer from 1-8, preferably from 1-4.

Among these monofunctional monomers, isobornyl (meth)acrylate, lauryl (meth)acrylate, and phenoxyethyl (meth)acrylate are particularly preferable.

Given as examples of commercially available products of the monofunctional monomers are Aronix M-101, M-102, M-111, M-113, M-117, M-152, TO-1210 (manufactured by Toagosei Chemical Industry Co., Ltd.), KAYARAD TC-110S, R-564, R-128H (manufactured by Nippon Kayaku Co., Ltd.), Viscoat 192, Viscoat 220, Viscoat 2311HP, Viscoat 2000, Viscoat 2100, Viscoat 2150, Viscoat 8F, Viscoat 17F (manufactured by Osaka Organic Chemical Industry Co., Ltd.)

Given as preferable examples of the polyfunctional monomers used as the component (C) are ethylene glycol di(meth)acrylate, dicyclopentenyl di(meth)acrylate, triethylene glycol diacrylate, tetra ethylene glycol di(meth)acrylate,

tricyclodecanediyl dimethylene di(meth)acrylate,  
tris(2-hydroxyethyl) isocyanurate di(meth)acrylate,  
tris(2-hydroxyethyl) isocyanurate tri(meth)acrylate,  
caprolactone modified tris(2-hydroxyethyl) isocyanurate  
5 tri(meth)acrylate, trimethylolpropane  
tri(meth)acrylate, ethylene oxide (hereinafter called  
[EO]) modified trimethylolpropane tri(meth)acrylate,  
propylene oxide (hereinafter called as [PO] as the case  
may be) modified trimethylolpropane tri(meth)acrylate,  
10 tripropylene glycol di(meth)acrylate, neopentyl glycol  
di(meth)acrylate, both terminal (meth)acrylic acid  
adduct of bisphenol A diglycidyl ether, 1,4-butanediol  
di(meth)acrylate, 1,6-hexanediol di(meth)acrylate,  
pentaerythritol tri(meth)acrylate, pentaerythritol  
15 tetra(meth)acrylate, polyester di(meth)acrylate,  
polyethylene glycol di(meth)acrylate, dipentaerythritol  
hexa(meth)acrylate, dipentaerythritol  
penta(meth)acrylate, dipentaerythritol  
tetra(meth)acrylate, caprolactone modified  
20 dipentaerythritol hexa(meth)acrylate, caprolactone  
modified dipentaerythritol penta(meth)acrylate,  
ditrimethylolpropane tetra(meth)acrylate, EO modified  
bisphenol A di(meth)acrylate, PO modified bisphenol A  
di(meth)acrylate, EO modified hydrogenated bisphenol A  
25 di(meth)acrylate, PO modified hydrogenated bisphenol A  
di(meth)acrylate, EO modified bisphenol F  
di(meth)acrylate, and phenol novolac polyglycidyl  
ether.

Given as commercially available products of  
30 these polyfunctional monomers are SA1002 (manufactured  
by Mitsubishi Chemical Co., Ltd.), Viscoat 195, Viscoat  
230, Viscoat 260, Viscoat 215, Viscoat 310, Viscoat  
214HP, Viscoat 295, Viscoat 300, Viscoat 360, Viscoat  
GPT, Viscoat 400, Viscoat 700, Viscoat 540, Viscoat  
35 3000, Viscoat 3700 (manufactured by Osaka Organic  
Chemical Industry Co., Ltd.), KAYARAD R-526, HDDA,  
NPGDA, TPGDA, MANDA, R-551, R-712, R-604, R-684,

- PET-30, GPO-303, TMPTA, THE-330, DPHA, DPHA-2H,  
DPHA-2C, DPHA-21, D-310, D-330, DPCA-20, DPCA-30,  
DPCA-60, DPCA-120, DN-0075, DN-2475, T-1420, T-2020,  
T-2040, TPA-320, TPA-330, RP-1040, RP-2040, R-011,  
5 R-300, R-205 (manufactured by Nippon Kayaku Co., Ltd.),  
Aronix M-210, M-220, M-233, M-240, M-215, M-305, M-309,  
M-310, M-315, M-325, M-400, M-6200, M-6400  
(manufactured by Toagosei Chemical Industry Co., Ltd.),  
Light acrylate BP-4EA, BP-4PA, BP-2EA, BP-2PA, DCP-A  
10 (manufactured by Kyoeisha Chemical Industry Co., Ltd.),  
New Frontier BPE-4, TEICA, BR-42M, GX-8345  
(manufactured by Daiichi Kogyo Seiyaku Co., Ltd.),  
ASF-400 (manufactured by Nippon Steel Chemical Co.,  
Ltd.), Ripoxy SP-1506, SP-1507, SP-1509, VR-77,  
15 SP-4010, SP-4060 (manufactured by Showa Highpolymer  
Co., Ltd.), and NK Ester A-BPE-4 (manufactured by  
Shin-Nakamura Chemical Industry Co., Ltd.).

The component (C) which is a constituent of  
the photo-curable resin composition of the present  
20 invention contains more than about 60% by weight of the  
polyfunctional monomer having three or more  
ethylenically unsaturated bonds in one molecule. The  
proportion of the polyfunctional monomer having three  
or more functional groups is preferably more than about  
25 70% by weight and more preferably more than about 80%  
by weight. A particularly preferable proportion of the  
polyfunctional monomer is substantially 100% by weight.  
If the proportion of the polyfunctional monomer having  
three or more functional groups is less than about 60%  
30 by weight, the photo-curing characteristic of the  
resulting resin composition is insufficient and the  
three-dimensional object fabricated tends to deform  
with time.

The polyfunctional monomer having three or more  
35 polyfunctional groups can be selected from the  
above-mentioned tri(meth)acrylate compounds,  
tetra(meth)acrylate compounds, penta(meth)acrylate



compounds, and hexa(meth)acrylate compounds. Among these, trimethylol propane tri(meth)acrylate, EO modified trimethylolpropane tri(meth)acrylate, dipentaerythritol hexa(meth)acrylate, dipentaerythritol  
5 penta(meth)acrylate, and ditrimethylolpropane tetra(meth)acrylate are particularly preferable. The above-mentioned monofunctional monomers and polyfunctional monomers can be used as the component (C) either individually or in combinations of two or  
10 more.

The proportion of the monofunctional monomer and polyfunctional monomer in the photo-curable resin composition of the present invention is generally about 5-30% by weight, preferably about 7-25% by weight, and  
15 more preferably about 10-20% by weight. If the proportion of the component (C) is too low, the photo-curing characteristic of the resin composition is insufficient. No three-dimensional object having sufficient mechanical strength can be fabricated from  
20 such resin composition. On the other hand, if the proportion of the composition (C) is too high, the resulting resin composition shrinks easily during by photo-curing and the mechanical strength such as toughness of the three-dimensional object tends to be  
25 reduced.

#### <Radically photo-initiator>

The radical photo-initiator (hereinafter called [component (D)]) which is a constituent of the  
30 photo-curable resin composition of the present invention is a compound which decomposes and generates radicals by receiving radiation such as light and initiates a radical polymerization reaction of the component (C) by the action of the radicals.

35 Given as specific examples of the radical photo-initiators which can be used as the component (D) are acetophenone, acetophenone benzyl ketal,

anthraquinone, 1-(4-isopropylphenyl)-2-hydroxy-  
2-methylpropan-1-one, carbazole, xanthone,  
4-chlorobenzo-phenone, 4,4'-diaminobenzophenone,  
1,1-dimethoxydeoxybenzoin, 3,3'-dimethyl-4-  
5 methoxybenzophenone, thioxanethone compounds,  
2-methyl-1-4-(methylthio) phenyl-2-morpholino-propane-  
2-on, 2-benzyl-2-dimethylamino-1-  
(4-morpholinophenyl)-butan-1-one, triphenylamine,  
2,4,6-trimethylbenzoyl diphenylphosphine oxides, bis  
10 (2,6-dimethoxybenzoyl)-2,4,4-tri-methylpentyl-phosphine  
oxide, benzyl dimethyl ketal, 1-hydroxycyclohexyl  
phenyl ketone, 2-hydroxy-2-methyl-1-  
phenylpropane-1-one, fluorenone, fluorene,  
benzaldehyde, benzoin ethyl ether, benzoin propyl  
15 ether, benzophenone, Michler's ketone,  
3-methylacetophenone, 3,3', 4,4'-tetra (t-butyl  
peroxycarbonyl) benzophenone (BTTB), and combined  
compositions of BTTB and xanthene, thioxanthene,  
cumarin, ketocumarin or other coloring matter  
20 photosensitizer. Among these, benzyl dimethyl ketal,  
1-hydroxycyclohexyl phenyl ketone,  
2,4,6-trimethylbenzoyl diphenylphosphine oxide,  
2-benzyl-2-dimethylamino-1-(4-morpholinophenyl)-butan-1  
-one, and the like are particularly preferred.

25           The above-mentioned radical photo-initiators  
can be used as the component (D) either individually or  
in combinations of two or more.

          The proportion of the component (D) in the  
photo-curable resin composition of the present  
30 invention is usually about 0.01-10% by weight and  
preferably about 0.1-8% by weight. If the proportion of  
the component (D) is too low, the rate (curing rate) of  
the radical polymerization is so reduced that  
fabricating time may be extended and the resolution may  
35 tend to be lower. On the other hand, if the proportion  
of the component (D) is too high, the excess  
photo-initiator sometimes has an adverse effect on the

curing characteristics of the resin composition and the specification, heat resistance, and handling of the three-dimensional object obtained from the resin composition.

5

<Polyether polyol>

The polyether polyol (hereinafter called [component (E)]) is an essential component for developing the photo-curability of the resin composition and the shape stability (resistance to deformation with time) of the three-dimensional object obtained from the resin composition. The polyether polyol used as the component (E) has on average about three or more, preferably about 3-6 hydroxyl groups in the molecules. If polyether polyols having less than three hydroxyl groups are used (e.g. polyether diols), the aim of developing the photo-curing characteristic can not be achieved and a three-dimensional object with sufficient mechanical strength can not be produced. On the other hand, if polyether polyols having on average 7 or more hydroxyl groups are used, the extensibility and toughness of the three-dimensional object obtained from the resin composition tends to be lower.

Given as specific examples of the component (E) are polyether polyols prepared by modifying polyhydric alcohol of more than 3 valences such as trimethylolpropane, glycerol, pentaerythritol, sorbitol, sucrose, polyether polyols or the like by a cyclic ether compound such as ethylene oxide (EO), propylene oxide (PO), butylene oxide, tetrahydrofuran, and the like. Specific examples of such polyether polyols are EO modified trimethylolpropane, PO modified trimethylolpropane, tetrahydrofuran modified trimethylolpropane, EO modified glycerol, PO modified glycerol, tetrahydrofuran modified glycerol, EO modified pentaerythritol, PO modified pentaerythritol, tetrahydrofuran modified pentaerythritol, EO modified

sorbitol, PO modified sorbitol, EO modified sucrose, PO modified sucrose, EO modified quodor. Among these, EO modified trimethylolpropane, PO modified trimethylolpropane, PO modified glycerol, PO modified sorbitol are desirable.

The preferable molecular weight of the polyether polyol is about 100 to about 2,000, and more preferably about 160-1,000. If polyether polyols having too small molecular weight are used as the component (E), it is difficult to prepare a three-dimensional object with stable in shape and properties. On the other hand, If polyether polyols having too large molecular weight are used as the component (E), the viscosity of the resulting resin composition becomes too large, resulting in lowering of the mechanical strength of the three-dimensional object obtained by the photo-fabrication process.

Given as specific examples of the polyether polyols which can be used as the component (E) are Sunnix TP-400, Sunnix GP-600, Sunnix GP-1000, Sunnix SP-750, Sunnix GP-250, Sunnix GP-400, Sunnix GP-600 (manufactured by Sanyo Chemical Co., Ltd.), TMP-3 Glycol, PNT-4 Glycol, EDA-P-4, EDA-P-8 (manufactured by Nippon Nyukazai Co., Ltd.), and G-300, G-400, G-700, T-400, EDP-450, SP-600, SC-800 (manufactured by Asahi Denka Kogyo Co., Ltd.).

The above-mentioned polyether polyols can be used as the component (E) either individually or in combinations of two or more.

The proportion of the component (E) in the photo-curable resin composition of the present invention is usually about 5-30% by weight, preferably about 7-25% by weight, and more preferably about 10-20% by weight. If the proportion of the component (E) is too low, the aim of developing the photo-curing characteristic can not be achieved and there are cases where a three-dimensional object with sufficient

stability in shape and properties can not be produced from the resin composition. On the other hand, if the proportion of the component (E) is too high, the photo-curing characteristics of the resin composition are degraded and there is a tendency for the mechanical strength of the three-dimensional object obtained from the resin composition to be reduced.

<Optional components>

10 ~~In addition to the above-mentioned components~~  
(A) to (E), various other components may be incorporated into the photo-curable resin composition of the present invention to the extent that the curability of this composition is not adversely  
15 affected. Such optional components include photosensitizers (polymerization promoters) of amine compounds such as triethanolamine, methyl diethanolamine, triethylamine, diethylamine; photosensitizers including thioxantone or its  
20 derivatives, anthraquinone or its derivatives, anthracene or its derivatives, perillene and its derivatives, benzophenone, benzoin isopropylether, and the like; and reaction diluents such as vinyl ether, vinyl sulfide, vinyl urethane, or vinyl urea.  
25 Also, various additives may be incorporated into the photo-curable resin composition. Such additives include polymers or oligomers, such as epoxy resin, polyamide, polyamideimide, polyurethane, polybutadiene, polychloroprene, polyether, polyester,  
30 styrene-butadiene-styrene block copolymer, petroleum resin, xylene resin, ketone resin, cellulose resin, fluorine oligomer, silicon oligomer, and polysulfide oligomer; polymerization inhibitors such as phenothiazine or 2,6-di-t-butyl-4-methyl phenol,  
35 polymerization initiation adjuvants, leveling agents, wettability improvers, surfactants, plasticizers, UV absorbers, silane coupling agents, inorganic fillers,

→

resin particles, pigments, dyes and the like.

The photo-curable resin composition of the present invention can be manufactured by homogeneously blending the above-mentioned components (A) to (E), the optional  
5 components, and the various additives.

It is desirable for the photo-curable resin composition of the present invention to possess a viscosity at 25°C in the range of about 50-1,000 cps, preferably about 70-500 cps.

10

<Photo-fabricating process>

The photo-curable resin composition of the present invention prepared in this manner is suitable as a photo-curable (liquid) material used in  
15 photo-fabrication processes. Specifically, a three-dimensional object with a desired shape can be obtained by using the photo-fabrication process, including selectively irradiating visible light, ultraviolet light, or infrared light on the  
20 photo-curable resin composition of the present invention, and feeding the energy required to cure the resin composition.

Various means may be used to selectively irradiate the photo-curable resin composition with  
25 light with no specific limitations. Such light irradiation means include, for example, a laser beam, a means for irradiating the composition with light and the like converged by a lens, mirror, or the like, while scanning, a means irradiating the composition  
30 with non-convergent light through a mask provided with a fixed pattern through which light is transmitted, and a means for irradiating the composition with light via a number of optical fibers bundled in a light  
35 the means using a mask, a mask electrooptically produces a mask image consisting of a light transmitting area and non-light-transmitting area

according to a prescribed pattern by the same theory as that of the liquid crystal display apparatus. A means using a scanning laser beam with a small spot size is preferred for selectively irradiating the resin composition with light, when a resulting three-dimensional object possesses minute parts or when high dimensional accuracy is required to form the three-dimensional object.

In the above means, the irradiated surface (for example, the plane scanned by light) of the resin composition placed in a vessel is either the liquid level of the resin composition or the surface at which the liquid contacts the wall of a transparent vessel. When the irradiated surface is the liquid level or the contact surface of the wall of the vessel, the light can be shone directly out of the vessel or through the vessel.

In the above photo-fabrication process, a desired solid shape can be made by curing fixed parts of the resin composition and then moving the light spot from the cured parts to the uncured parts continuously or gradually to laminate the cured parts. There are various methods for moving the light spot, for example, a method for moving any of the light source, the vessel for the resin composition, or the cured parts of the resin composition. Also there is a method in which fresh resin composition is supplied to the cured resin composition in the vessel.

In a typical photo-fabrication process, the surface of a support stage capable of being elevated in the vessel is lowered quite a bit from the liquid surface to form a thin layer (1) of the resin composition, the thin layer (1) is selectively irradiated with light to form a solid cured resin layer (1'). The resin composition is supplied over this thin layer (1') to form a second thin layer (2), and this thin layer (2) is selectively irradiated with light to

laminate a new solid cured resin layer (2') on the thin layer (1'). This step is repeated for a prescribed number of times, with or without changing the pattern subjected to light irradiation, to produce a  
5 three-dimensional object consisting of a multiple number of cured resin layers (n') which are integrally laminated.

The three-dimensional object fabricated in this manner is discharged from the vessel and processed  
10 to remove the unreacted photo-curable resin composition remaining on the surface, and washed by a solvent, as required. Given as examples of solvents are an organic solvent which is represented by an alcohol, such as isopropyl alcohol or ethyl alcohol, an organic solvent  
15 such as acetone, ethyl acetate, methylethyl ketone, an aliphatic organic solvent such as a terpene, or a low viscosity liquid thermosetting resin or photo-curable resin.

When forming a three-dimensional object with  
20 a smooth surface, it is desirable that the cured product be washed using the thermosetting resin or photo-curable resin. In this case, it is necessary to post-cure the product by heat emission or light irradiation depending on the type of solvent used in  
25 the washing stage. This post-curing treatment is effective not only for curing the resin remaining uncured on the surface of the laminated body, but also for curing the resin composition which remains uncured inside the laminated body. Thus, the post-curing  
30 treatment is also effective in the case where the fabricated three-dimensional object is washed with an organic solvent.

The three-dimensional object obtained in this manner has high mechanical strength, high dimensional  
35 accuracy, and excellent heat resistance. Also, the three-dimensional object exhibits high stability in maintaining a fixed shape and lasting stable



properties. Therefore, the three-dimensional object prepared from the resin composition is preferably used for trial products for mechanical parts.

It is desirable to cover the surface of the  
5 three-dimensional object by a heat-curable or  
photo-curable hard coating agent to improve the  
strength and heat resistance of the surface. As such a  
hard coating agent, an organic coating agent such as  
acryl resin, epoxy resin, silicon resin, or the like,  
10 or an inorganic coating agent can be used. These hard  
coating agents may be used individually or in  
combination of two or more.

#### EXAMPLES

15 The present invention will be explained in  
more detail by way of examples, which are not intended  
to be limiting of the present invention.

#### <Example 1>

20 According to the formulation shown in Table  
1, 48% by weight of  
3,4-epoxycyclohexylmethyl-3',4'-epoxycyclohexane  
carboxylate (UVR-6110, manufactured by Union Carbide),  
16% by weight of butanedioldiglycidyl ether (KRM-2720,  
25 manufactured by Asahi Denka Kogyo), 5% by weight of  
trimethylolpropane triacrylate (Viscoat 295,  
manufactured by Osaka Organic Chemical), 5% by weight  
of dipentaerythritol hexaacrylate (KAYARAD DPHA,  
manufactured by Nippon Kayaku), 5% by weight of  
30 bisphenol A diglycidylether diacrylate (Novacure 3700,  
manufactured by Radcure), 2% by weight of  
triallylsulfoniumhexafluoro antimonate (UVI-6974,  
manufactured by Union Carbide), 0.2% by weight of  
diethylthioxantone (DETX-S, manufactured by Nippon  
35 Kayaku), 2% by weight of 1-hydroxycyclohexyl-  
phenylketone (Irgacure 184, manufactured by Ciba  
Geigy), and 17% by weight of PO modified glycerol

(Sunnix GP-250, manufactured by Sanyo Chemical) were placed in a vessel equipped with a stirrer and the mixture was reacted with stirring at 60°C for one hour to prepare a transparent liquid composition (the resin composition of the present invention). The viscosity of the liquid composition resulting was measured at 25°C by means of a B type viscometer. The viscosity of the composition was 150 cps.

10 <Examples 2-9>

Transparent liquid compositions (the photo-curable resin compositions of the present invention) were prepared in the same manner as in Example 1 according to the formulations shown in Table 1, except that the different components were used. The viscosity (at 25°C) of the liquid compositions prepared were each measured by a B-type viscometer. The results are shown in Table 1.

20 <Comparative Examples A-D>

Transparent liquid compositions (comparative photo-curable resin compositions) were prepared in the same manner as in Example 1 according to the formulations shown in Table 1, except that the different components and optional components were used. The viscosity (at 25°C) of the prepared liquid compositions were each measured by a B-type viscometer. The results are shown in Table 1.

TABLE 1

Amounts in wt. %	EXAMPLE										COMPARATIVE EXAMPLE			
	1	2	3	4	5	6	7	8	9		A	B	C	D
Component A														
- 3,4-Epoxy-cyclohexyl-methyl-3',4'-epoxy-cyclohexane carboxylate	48	48	48	48	48	48	48	48	48		48	48	48	48
- Butanediol diglycidyl ether	16	18	18	18	18	16	18	16	18		18	18	18	18
Component B														
Triarylsulfonium hexafluoroantimonate														
- UVI-6974*1	2	2	-	-	2	2	1	2	2		2	1	2	2
- UVI-6970*1	-	-	2	-	-	-	-	-	-		-	-	-	-
- SP-171*2	-	-	-	2	-	-	-	-	-		-	-	-	-
Component C														
- Trimethylolpropane triacrylate	5	8	8	8	8	8	8	5	8		8	-	-	8
- Dipentaerythritol hexaacrylate	5	5	5	5	5	5	5	5	5		5	4	4	5
- Bisphenol A diglycidyl ether diacrylate	5	-	-	-	-	-	-	5	-		-	8	8	-
Component (D)														
1-Hydroxycyclohexyl phenyl ketone	2	2	2	2	2	2	1	2	2		2	1	2	2
Component (E)														
- PO-modified glycerine	17	17	17	17	-	-	-	17	-		-	-	17	-
- PO-modified trimethylol-propane*3	-	-	-	-	12	-	-	-	-		-	-	-	-
- PO-modified trimethylol-propane*4	-	-	-	-	-	17	-	-	-		-	-	-	-
- PO-modified sorbitol*5	-	-	-	-	-	-	19	-	-		-	-	-	-
- Polyether polyol*6	-	-	-	-	-	-	-	-	17		-	-	-	-

5

10

15

20

25

Amounts in wt. %	EXAMPLES									COMPARATIVE EXAMPLE			
	1	2	3	4	5	6	7	8	9	A	B	C	D
Optional Component													
- Diethylthioxanetone*7	0.2	-	-	-	-	-	-	-	-	-	-	-	-
- 1,4-Butenediol adipic acid ester*8	-	-	-	-	5	-	-	-	-	-	-	-	17
Caprolactone triol*9	-	-	-	-	-	-	-	-	-	17	20	-	-
Viscosity (25°C) (cps)	150	110	110	110	150	100	250	150	170	160	220	160	220

5

\*1) [UVI-6974], [UVI-6970] (manufactured by Union Carbide Corporation)

\*2) [SP-171] (manufactured by Asahi Denka Industries Co., Ltd.)

\*3) [Sunnix GP-400] (manufactured by Sanyo Chemical Co., Ltd.)

\*4) [TMP-30U] (manufactured by Nippon Nyukazai Kogyo Co., Ltd.)

\*5) [Sunnix SP-750] (manufactured by Sanyo Chemical CO., Ltd.)

\*6) [Desmophenon 1915] (manufactured by Bayer )

\*7) A photosensitizer

\*8) [Nipporan 4009] (manufactured by Nippon Polyurethane Industries Co., Ltd.)

\*9) [TONE-0301] (manufactured by Union Carbide Corporation)

10

15

As shown in Table 1, all the compositions prepared in Examples 1 to 9 have an optimum viscosity as the resin composition used in the photo-fabrication process.

5   <Evaluation of photo-curable resin compositions>

          The photo-curable resin compositions prepared in Examples 1-9 and Comparative Examples A-D were evaluated by measuring the Young's modulus of the cured products and the deformation rate with time. The  
10   results are shown in Table 2.

<Change in Young's modulus with time>

          (1) Preparation of test specimens

A coated film with a thickness of 200  $\mu\text{m}$  was prepared  
15   by applying a composition to a glass plate using an applicator. The surface of the film was irradiated with ultraviolet light at a dose of 0.5 J/cm<sup>2</sup> using a conveyer curing apparatus equipped with a metal halide lamp to prepare a half-cured resin film. Next, the half  
20   cured resin film was peeled from the glass plate and placed on releasable paper. The side opposite to that first cured by irradiation was irradiated with ultraviolet light at a dose of 0.5 J/cm<sup>2</sup> to prepare a completely cured resin film.

25           The cured resin film was allowed to stand under the atmospheric conditions described below to produce three test specimens (1) to (3).

          Specimen (1) was prepared at a temperature of 23°C, under a humidity of 50% and allowed to stand in a  
30   thermo-hygrostat for 24 hours.

          Specimen (2) was prepared at a temperature of 23°C, under a humidity of 50% and allowed to stand in a thermo-hygrostat for 30 days.

          Specimen (3) was prepared at a temperature of  
35   23°C, under a humidity of 80% and allowed to stand in a thermo-hygrostat for 30 days.

          (2) Measurement of Young's modulus

The Young's modulus of the test specimen was measured at a temperature of 23°C under a humidity of 50% under the conditions of a drawing rate of 1 mm/min and a bench mark distance of 25 mm.

5

[Rate of deformation with time]

(1) Preparation of test specimen

Using an photo-fabrication apparatus (Solid Creator JSC-2000, manufactured by Sony Corporation) equipped  
10 with a light source for applying an argon ion laser beam (wave length of 351 nm and 365 nm), a photo-curable resin composition was selectively irradiated by the laser beam under the conditions of a laser power of 100 mW measured on the irradiated  
15 surface (liquid level) and a scanning speed of 400 mm/s to form a cured resin film with a thickness of 0.15 mm. This fabricating procedure was repeated many times to form a model for measurement (hereinafter called [camber model]) as shown in Fig. 1. The camber model  
20 was then discharged from the photo-fabrication apparatus, after which the residual resin composition adhering to the outer surface of the camber model was released, and the excess resin composition was removed using a terpene-type solvent.

25 (2) Measurement

As shown in Fig.1, measurements were made by securing a foot part 11 of a camber model 10 to a plane 20 and measuring the distance (lift length  $\Delta h$  (mm)) between the plane 20 and the bottom of a foot part 12 to  
30 determine the primary camber rate.

After the camber model 10 was allowed to stand in a thermo-hygrostat at 23°C under a humidity of 50% for 3 days, the camber rate was measured in the same manner as in the above procedure.

35

TABLE 2

	Changes with time of Young's modulus (kg/mm <sup>2</sup> )			Changes with time	
	Specimen (1)	Specimen (2)	Specimen (3)	Warping (mm)	
	Temp., humidity, time period allowed to stand			Initial value	After 30 days
	23°C, 50%, 24Hrs			23°C, 80%, 30Days	
Example					
1	150	145	140	0.00	0.10
2	170	170	160	0.00	0.02
3	170	170	160	0.00	0.02
4	170	170	160	0.00	0.02
5	155	140	140	0.02	0.09
6	160	170	155	0.00	0.04
7	185	190	170	0.00	0.05
8	150	135	120	0.01	0.15
9	130	110	100	0.04	0.12

5

10

15

	Changes with time of Young's modulus (kg/mm <sup>2</sup> )				Changes with time	
	Specimen (1)	Specimen (2)	Specimen (3)		Warpling (mm)	
	Temp., humidity, time period allowed to stand				Initial value	After 30 days
	23°C, 50%, 24Hrs	23°C, 50%, 30Days	23°C, 80%, 30Days			
Comparative Example						
A	165	110	70		0.02	0.42
B	140	70	10		0.00	0.58
C	130	80	60		0.02	0.22
D	120	85	35		0.10	0.22



As is clear from Table 2, all cured products from the photo-curable resin compositions of the present invention prepared in the Examples 1-9, especially Examples 1-7 had excellent mechanical properties with a large Young's modulus and a lasting small chamber rate.

In contrast, the cured product formed from the composition prepared in the Comparative Example A, using caprolactone modified triol (polyester polyol) instead of the component (E) exhibited a large change in the Young's modulus and minimal deformation with time.

Evaluations were made for the cured products formed from the following compositions prepared from Comparative Examples: the composition prepared in Comparative Example B, using caprolactone modified triol instead of the component (E) and containing less than 60% by weight of a poly functional monomer having three or more polyfunctional groups in the component (C); the composition prepared in Comparative Example C, containing less than 60% by weight of a polyfunctional monomer having three or more polyfunctional groups in the component (C), although the composition contained the component (E); and the composition prepared in Comparative Example D, using adipic ester of 1,4 butane diol instead of the component (E). All these cured products from the compositions prepared from the Comparative Examples had low mechanical strength with a small Young's modulus, and a large change in Young's modulus and camber rate with time.

The photo-curable resin composition of the present invention can produce cured products having little change in mechanical properties and deformation rate with time, maintaining high dimensional accuracy, and exhibiting excellent mechanical strength and heat resistance. The cured product can therefore be suitably used as a three-dimensional object, for example, a trial product for mechanical parts.

In addition, the photo-fabrication process using the photo-curable resin composition of the present invention can provide a three-dimensional object possessing excellent primary characteristics and high durability.

#### BRIEF DESCRIPTION OF THE DRAWINGS

Figure 1 is a diagram showing the configuration of the model for the measurement of deformation rate with time and explaining measuring method.

#### DESCRIPTION OF SYMBOLS

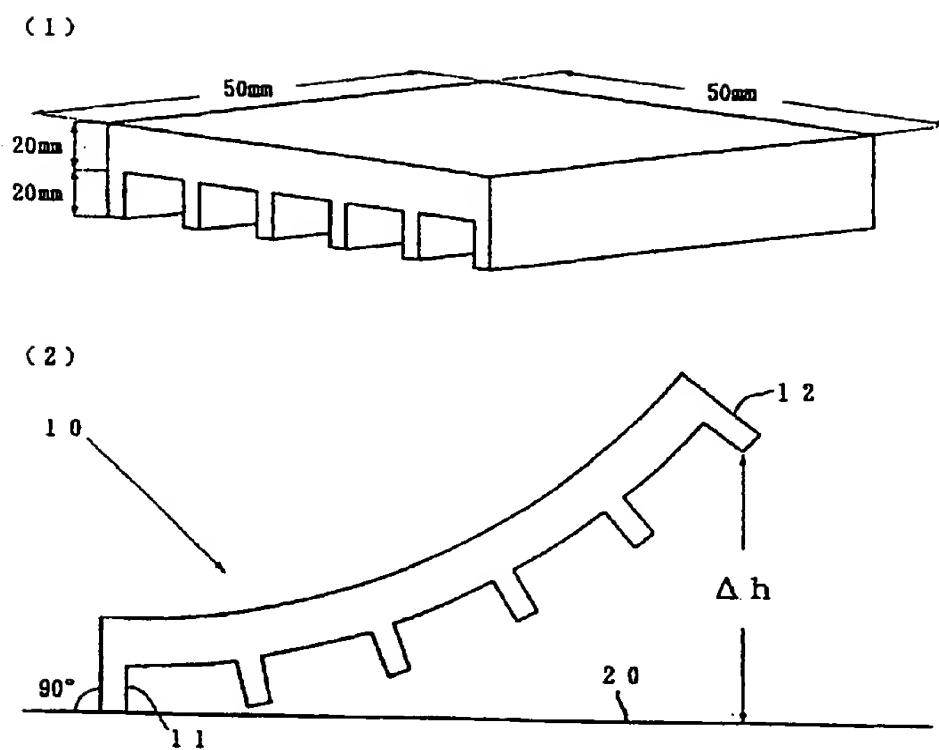
10 Camber model  
15 11, 12 Foot part  
20 Plane

## WHAT IS CLAIMED IS:

1. A photo-curable resin composition comprising,  
5 (A) a cationic polymerizable organic compound,  
(B) a cationic photo-initiator,  
(C) an ethylenically unsaturated monomer,  
(D) a radical photo-initiator, and  
(E) a polyether polyol having on average about  
10 three or more hydroxyl groups, wherein at least  
about 60% by weight of the said ethylenically  
unsaturated monomer is a polyfunctional monomer  
having three or more ethylenically unsaturated  
bonds in one molecule.
- 15 2. A photo-curable resin composition according to  
claim 1, wherein the composition has been  
formulated from components comprising:  
about 30-85 wt.% of component (A)  
about 0.1-10 wt.% of component (B)  
20 about 5-30 wt.% of component (C)  
about 0.1-10 wt.% of component (D)  
about 5-30 wt.% of component (E)
3. A photo-curable resin composition according to  
claim 1 wherein the composition has a viscosity at  
25 25°C in the range of about 50-1000 cps.
4. A photo-curable composition according to claim 1,  
wherein component (A) is a compound comprising at  
least one epoxy group.
5. A photo-curable composition according to claim 1,  
30 wherein component (C) is a compound comprising at  
least one (meth)acrylate group.
6. A photo-curable composition according to claim 1,  
wherein component (E) has on average about 3-6  
hydroxyl groups.
- 35 7. A photo-curable composition according to claim 1,  
wherein component (E) has a molecular weight of  
about 100-2,000.

8. Process for the fabrication of a three-dimensional object comprising the steps of forming a curable resin layer having a specified pattern by selectively irradiating with light feeding the  
5 photo-curable resin composition equivalent to one layer to form another thin layer of the composition over this cured resin layer, and selectively irradiating this thin layer with light to form a new cured resin layer which is  
10 integrally laminated over the previously formed cured resin layer and repeating this step a number of times, with or without changing the pattern in which the light is irradiated to form a three-dimensional object consisting of integrally  
15 laminated multiple cured resin layers, wherein a resin composition according to any one of claims 1-7 is used.

Figure 1



# INTERNATIONAL SEARCH REPORT

Int. onal Application No

PCT/NL 97/00177

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 G03C9/08 G03F7/029

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 G03C G03F

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 605 361 A (CIBA-GEIGY AG) 6 July 1994 cited in the application see page 8, line 55 - line 57 see page 11; examples 4,5 see page 11; claims; examples 12,13 ---	1-8
A	EP 0 535 828 A (IMPERIAL CHEMICAL INDUSTRIES PLC) 7 April 1993 see page 9; example 9 -----	1-8

☐ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents:

- "A" document defining the general state of the art which is not considered to be of particular relevance
- "E" earlier document but published on or after the international filing date
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Date of the actual completion of the international search

8 July 1997

Date of mailing of the international search report

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# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/NL 97/00177

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